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SEARCH FOCUSSED ON UNLY THE ANOWN

Regioselective oxidations of equilenin derivatives catalyzed by a rhodium(III) porphyrin complex-contrast with the manganese(III) porphyrin Yang, Jerry; Breslow, Ronald
Department of Chemistry, Columbia University, New York, NY, 10027, USA
Tetrahedron Letters (2000), 41(42), 8063-8067
CODEN: TELEAY; ISSN: 0040-4039 SOS

Elsevier Science Ltd Journal

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Catalytic reactions of metalloporphyrins. 3. Catalytic modification of Mydroboration-oxidation of olefins with rhodium(III) porphyrin as catalyst Aoyama, Yasuhiro; Tanaka, Yasutaka; Fujisawa, Takeshi; Watanabe, Takamichi; Toi, Hiroo; Ogoshi, Hisanobu Dep, Mater. Sci. Technol., Hechnol. Univ. Nagaoka, Nagaoka, 940-21, Japan CODEN: JOCEAH; ISSN: 0022-3263 Efficient olefin oxygenation with tetrahydroborate and dioxygen catalyzed by a rhodium porphyrin complex Acyama, Yasuhiro; Watanabe, Takamichi; Onda, Hiroyuki; Ogoshi, Hisanobu Dep. Mater. Sci., Technol. Univ. Nagaoka, Niigata, 949, Japan Tetrahedron Letters (1981), 24(11), 1183-6 CODEN: TELEAY; ISSN: 0040-4039 Pevelogment of supramolecular metalloprotein mimics
Feiters, M. C.; Gebbink, R. J. M. Klein; Schenning, A. P. H. J.; van
Strijdonck, G. P. F.; Martens, C. F.; Nollet, R. J. M.
Dep. Org. Chem., Univ. Nijmegen, Nijmegen, 6555 ED, Neth.
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CODEN: PACHAS; ISSN: 0033-4545 Selective electrooxidation of carbon monoxide with carbon-supported rhodium and iridium porphyrins at low potentials in acid electrolyte Van Baar, J. F.; Van Veen, J. A. R.; De Wit, N. K./Shell-Lab., Shell Res. B. V., Amsterdam, Neth. Electrochimica Acta (1982), 27(1), 57-9

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Equilibrium thermodynamic studies for reactions of rhodium
Equilibrium thermodynamic studies for reactions of rhodium
Porphyria hydride with CO, aldehydes and olefins in aqueous media
Fu, Xuefeng; Wayland, Bradford B.
Pepartment of Chemistry, University of Pennsylvania, philadelphia, PA,
19104, USA
Abstracts of Papers, 228th ACS National Meeting, Philadelphia, PA, United
States, August 22-26, 2004 (2004), INOR-600 Publisher: American Chemical
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Olefins in Water: Organo-Rhodium Porphyrin Bond
Dissociation Free Energies
Fu, Xuefeng; Wayland, Bradford B.
Department of Chemistry, University of Pennsylvania, Philadelphia, PA,
19104-6323, USA
JOURNEL Of American Chemical Society (2005), 127(47), 16460-16467
CODEN: JACSAT, ISSN: 0002-7863 Spectral properties of cationic water-soluble metalloporphyrins immobilized in a perfluorosulfonated ion-exchange membrane Vasil'ev, Victor V.; Borisov, Sergey M.; Maldotti, Andrea; Molinari, THERE ARE 48 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORWAT ANSWER 3 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN 2004:354167 CAPLUS 141:321728 (CONVENTIONAL OR CONVENTIONALS)
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Alessandra So Department of Chemistry, Russian State Pedagogical University, St. Petersburg, 191186, Russia So Journal of Porphyrins and Phthalocyanines CODEN: JPPHEZ: ISSN: 10884-246 CODEN: JPPHEZ: ISSN: 10884-246 B Society of Porphyrins 6 Phthalocyanines DT JOURNAL ALCITATIONS AVAILABLE FOR THIS RECORD ALCITATIONS AVAILABLE IN THE RE FORMAT 10.310322 TI SEALING COPYRIGHT 2006 ACS on STN AN 2004:87921 CAPLUS CAPLUS SO Journal of the American Chemical Society (2004), 126(8), 2623-2631 CODEN: JACSAT: ISSN: 0002-7863 PB American Chemical Society JOURNAL AN 2002:851347 CAPLUS COPYRIGHT 2006 ACS on STN ALCITATIONS AVAILABLE IN THE RE FORMAT 1.1 ANSWER SO 13 CAPLUS COPEN: JACSAT: SSN: 1002-7863 DN 199:111611 TI PORPHYRING With virucidal activity, and use in the treatment of sexually transmitted diseases IN Compans, Richard W.: Marzilli, Luigi G.: Sears, Amy E.: Dixon, Dabney W. Emory University, USA; Georgia State University Research Foundation, Inc. SO CODEN: PATENT NO. KIND DATE APPLICATION NO. DATE PATENT NO.	1716	L7 ANSWER 6 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN AN 2003:183713 CAPLUS T1 Aqueous organometallic reactions of rhodium porphyrins
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ANSWER 9 OF 13 CAPLUS COPYRIGHT 2006 ACS on STN Reactions of rhodium porphryins with small mol. substrates in water were performed and the equilibrium consts. for these reactions were determined

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tetra (4-sulfonatophenyl)porphyrin [(TSPP)Rh], reacts with dihydrogen, carbon monoxide and olefins in water to form the hydride, formyl and 9-hydroxy alkyl derivay. Results from these studies will be presented in the context of substrate reactions of **rhodium** porphyrins with dihydrogen, carbon monoxide and ethene in non-aqueous

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i.	<pre>14.1.001 Intramolecular amidation of sulfamates 1,2,3-oxathiazolidine-2,2-dione and tetrahyfor-1,2,3-oxathiazine-2,2-dione derivatives catalyzed by meralloworshyrins</pre>
IN PA SO TA	Che, Chi-Ming; Liang, Jiang-Lin Hong Kong U.S. Pat. Appl. Publ., 12 pp., Contin-part of U.S. Ser. No. 202,581. CODEN: USXXCO
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II	Intramolecular C-N Bond Formation Reactions Catalyzed by Ruthenium Porphyrins; Amidation of Sulfamate Esters and Aziridination of Unsaturated Sulfonamides
AU	Shi-Xue; Huang, Jie-Sheng; Che, Chi-Ming and Open Laboratory of Chemical Biology, Inst
	of Molecular Technology for Drug Discovery and Synthesis, University of Hong Kong, Hong Kong
So	Journal of Organic Chemistry (2004), 69(11), 3610-3619 CODEN: JOCEAH; ISSN: 0022-3263
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II Synthesis of meso-tetra-(3,5-dibromo-4-hydroxyphenyl)- porphyrin
and its application to second-derivative spectrophotometric determination
of lead in clinical samples
Li, Zaijun; Zhu, Zhengzhong; Tang, Jan; Pan, Jiaomai
CS Dep. Chem. Eng., Wuxi University of Light Industry, Wuxi, 214036, Peop.
Rep. China
CODEN: ANALO; ISSN: 0003-2654
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RE.CNT 30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD
TI Highly diastereo- and enantioselective intramolecular amidation of saturated C-+ bonds catalyzed by ruthenium porphyrina

AU Liang, Jiang-Lin; Yuan, Shi-Xue; Huang, Jie-Sheng; Yu, Ming-Yiu; Che, Chi-Ming
CS Department of Chemistry and Open Laboratory of Chemical Biology of the Institute of Molecular Technology for Drug Discovery and Synthesis, The University of Hong Kong, Hong Kong, Hong Kong, Angewandte Chemie, International Edition (2002), 41(18), 3465-3468

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CODEN: ACIETS, ISSN: 1433-881

BM Miley-VCH Verlag GmbH & Co. KGaA
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Hayashi, Takashi; Kato, Tomoki; Kaneko, Tomomasa; Asai, Tomohito; Ogoshi, Department of Synthetic Chemistry and Biological Chemistry, Faculty of Method for supporting metalloporphyrins on polybenzimidazole porous articles for catalysts DATE Carbene insertion into oxygen-hydrogen bonds by metalloporphyrin Engineering, Kyoto University, Sakyo-ku, Kyoto, 606-01, Japan Journal of Organometallic Chemistry (1994), 473(1-2), 323-7 CODEN: JORALF ISSN: 0022-328x Processes for producing carbamates and isocyanates Leung, Tak W.; Dombek, Bernard D. Union Carbide Chemicals and Plastics Technology Corp., USA APPLICATION NO. APPLICATION NO. US 1990-631962 US 1987-28353 ANSWER 1 OF 3 CAPLUS COPYRIGHT 2006 ACS on STN 1995:27111 CAPLUS 122:159797 ANSWER 2 OF 3 CAPLUS COPYRIGHT 2006 ACS on STN 1993:494707 CAPLUS 119:94707 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2006 ACS on STN 1989:464960 CAPLUS 19930316 19901221 19890124 19870320 (RHODIUM OR RHODIUMS) DATE DATE Shepherd, James P. Hoechst Celanese Corp., USA U.S., 10 pp. CODEN: USXXAM KIND KIND 3 L10 AND RHODIUM and rhodium 67191 RHODIUM 31 RHODIUMS 67192 RHODIUM English CASREACT 122:159797 CASREACT 119:94707 1990-631962 U.S., 13 pp. CODEN: USXXAM PI US 4800188 PRAI US 1987-28353 US 5194660 US 1990-631 PATENT NO. catalysts PATENT NO. 111:64960 Patent English Hisanobu English Journal => s 110 => d 1-3 COD DT Pat LA Eng S PI PRAI OS L111 DN DN TI L11 DN TI TI IN PA LIII AN DN TII IN PA SO SO DT LA EAN. SS FIRS

ANSWER 2 OF 3 CAPLUS COPYRIGHT 2006 ACS on STN Carbamates are prepared by oxidative carbonylation of primary or secondary amines or ureas with CO in presence of an alc., an O-containing oxidizing agent, metalloporphyrin or metal phthalocyanine catalyst derived from

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Meso-aryl substituted metalloporphyrins supported on imidazole propyl gel (IPG). Catalytic activity in the oxidation of cyclohexane and characterization of iron porphyrin-IPG systems lamamoto, Yassuko; Ciuffi, Katia Jorge; Sacco, Herica Cristina; Prado, Cynthia Maria C.; de Moraes, Margarida; Nascimento, Otaciro Rangel Journal of Molecular Catalysis (1994), 88(2), 167-76 CODEN: JMCABS; ISSN: 0304-5102 r, Group IIIa-Va and Group VIII metals, and an iodine-containing promoter. Decomposition of carbamates prepared in this manner affords isocyanates. reaction of 3.0 g tert-Bunkly. 0.20 g CoPc (Pc = phthalocyanine dianion), and 1.0 g NaI with 40 g EtOH under 80 psi 02/1520 psi CO afforded 99% yield of Et N-tert-Bu carbamate. Biomimetic catalyst development for natural gas conversion Showalter, Margaret C.; Shelnutt, John A.; Medforth, Craig J.; Quirke, Fuel Sci. Dep., Sandia Natl. Lab., Albuquerque, NM, 87185-0710, USA Preprints of Papers - American Chemical Society, Division of Fuel Chemistry (1994), 39(4), 1002-5 CODEN: ACFPAI, ISSN: 0569-3772 American Chemical Society, Division of Fuel Chemistry Anchored manganese and ruthenium porphyrins as catalysts in the decomposition of cyclohexyl hydroperoxide Hansen, C. B.; Hoogers, G. J.; Drenth, W. Debije Inst., Utrecht Univ., Utrecht, 3584 CH, Neth. Journal of Molecular Catalysis (1993), 79(1-3), 153-63 CODEN: JMCADS, ISSN: 0304-5102 (BETTER OR BETTERS)
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1	studies on the catalytic effects of organic compounds by polymer-bonded metalloporphyrins	
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S S	Dep. Chem., Yonsel Univ., Wonju, 222-701, S. Korea Journal of the Korean Chemical Society (1992), 36(5), 744-52	
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AN	1992:514005 CAPLUS	
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TI	Nitrated metalloporphyrins as catalysts for alkane oxidation	
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	R: BE, DE, FR, GB, IT, NL	
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PRAI	NI US 1991-758147 A2 19910912 IIS 1882-883166 A 18820602	
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AN	1991:655400 CAPLUS	
DN	115:255400	
TI	Highly oxidation resistant inorganic-porphyrin analog polyoxometalate	
	oxidation catalysts. 2. Catalysis of otelin epoxidation and aliphatic and aromatic hydroxylations starting from $\alpha 2$ -	
	P2W17061(Mn+·Br)(n-11) (Mn+ = Mn3+, Fe3+, Co2+, Ni2+, Cu2+), including	
AU	quantitative comparisons to metalloporphyrin catalysts Mansuv. Daniel: Bartoli. Jean Francois: Battioni. Pierrette: Ivon. David	
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S	Lab. Chim. Biochim. Pharmacol. Toxicol., Univ. Rene Descartes, Paris,	
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SO	Journal of the American Chemical Society (1991), 113(19), 7222-6	
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chlorides Ando, Wataru; Tajima, Rieko; Takata, Toshikazu Dep. Chem., Univ. Tsukuba, Ibaraki, 305, Japan Tetrahedron Letters (1982), 23(16), 1685-8 CODEN: TELEAY; ISSN: 0040-4039

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Sch. Chem. Sci., Univ. Illinois, Urbana, IL, 61801, USA Journal of the American Chemical Society (1986), 108(23), 7281-6 CODEN: JACSAT; ISSN: 0002-7863

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Synthesis, characterization and reactivity of novel bis(tosyl)imidoruthenium(VI) porphyrin complexes; x-ray crystal bis(tosyl)imidoruthenium(VI) porphyrin cyllanitorium of a tosylamidoruthenium(IV) porphyrin kustructure of a tosylamidoruthenium(IV) porphyrin kustructure of a tosylamidoruthenium(IV) porphyrin bus Shie-Ming chemistry, Cheng, Chengi, Chenistry, The University of Hong Kong, Hong Kong Communications (Cambridge) (1997), (17), 1655-1656
CODEN: CHCOFF: ISSN: 1359-7345
Royal Society of Chemistry
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The monopersulfate oxidation of 4-isopropylbenzoic acid performed in H2180
and catalyzed by a water-soluble metalloporphyrin indicated that half of the
oxygen atoms incorporated in 4-[1-hydroxy-1-methylethyl]benzoic acid, the
primary hydroxylation product, came from water. A redox tautomerism of
the active high-valent hydroxo-metal-oxo porphyrin intermediate
coupled with an oxygen rebound mechanism explained this result. Under
similar conditions, ketoprofen was directly oxidized to
3-benzoylacetophenone, via at least two different reaction pathways.
Trapping of radical intermediates by mol. oxygen competed with the oxygen
rebound mechanism.
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024detions Catalyzed by a Water-Soluble Metalloporphyrin
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Chemistry Department, University of Guelph, Ontario, ON, NIG 2M1, Can.
Inorganic Chemistry (1997), 36(16), 3488-3492
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American Chemical Society
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Liang, Jiang-Lin; Yuan, Shi-Xue; Huang, Jie-Sheng; Che, Chi-Ming Department of Chemistry and Open Laboratory of Chemical Biology, Institute of Molecular Technology for Drug Discovery and Synthesis, University of Hong Kong, Hong Kong Gorganic Chemistry (2004), 69(11), 3610-3619
CODEN: JOCENH; ISSN: 0022-3263
American Chemical Society Preparation of **cyclic sulfamidates** by metalloporphyrin-catalyzed oxidative intramolecular amidation of sulfamate 20030718 SE, MC, PT, HU, SK Department of Chemistry and Open Laboratory of Chemical Biology of the Institute of Molecular Technology for Drug Discovery and Synthesis, The University of Hong Kong, Hong Kong Angewandte Chemie, International Edition (2002), 41(18), 3465-3468 CODEN: ACIEFS; ISSN: 1433-7851 Highly diastereo- and enantioselective intramolecular amidation of saturated C-H bonds catalyzed by ruthenium **porphyrins** Liang, Jiang-Lin; Yuan, Shi-Xue; Huang, Jie-Sheng; Yu, Wing-Yiu; Che, Chi-Ming DATE Intramolecular C-N Bond Formation Reactions Catalyzed by Ruthenium Porphyrins: Amidation of Sulfamate Esters and Aziridination of Unsaturated Sulfonanides SO JOURNAL V. ... SSN: VV.L. COORDEN: JOSEAH; ISSN: VV.L. COORDEN: JOSEAH; ISSN: VV.L. COORDEN: JOSEAH; ISSN: VV.L. COORDEN: JOURNAL JOURNAL JOURNAL JOURNAL A ENGLISH OS CASRACT 141:123207
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asymmetric synthesis of (-)-aphanorphine Bower, John F.; Szeto, Peter; Gallagher, Timothy School of Chemistry, University of Bristol, Bristol, BSB 1TS, UK Chemical Communications (Cambridge, United Kingdom) (2005), (46), 5793-5795 THERE ARE 31 CITED REFERENCES AVAILABLE FOR THIS RECORD ALL CITATIONS AVAILABLE IN THE RE FORMAT ANSWER 1 OF 39 CAPLUS COPYRIGHT 2006 ACS on STN 2005:1245779 CAPLUS Cyclic sulfamidates as lactam precursors. An efficient CODEN: CHCOFS; ISSN: 1359-7345 Royal Society of Chemistry English IT 31 Journal PB Roy DT Jou LA Eng RE,CNT L22 AN TI So

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Homoserine-derived **cyclic sulfamidate** as chiral adduct cor the diversity-oriented synthesis of lactam-bridged dipeptides galaud, Fabrice; Lubell, William D. Departement de Chimie, Universite de Montreal, Montreal, QC, H3C 3J7, Can. Bipoplymers; (2005), 80(5), 665-674 CODEN: BIPWAA, ISSN: 0006-3525 John Wiley & Sons, Inc. L22 AN DN TI SO

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Convenient Enantioselective Synthesis of (S)- α -Trifluoromethylisoserine AU

Avenoza, Alberto; Busto, Jesus H.; Jimenez-Oses, Gonzalo; Peregrina, Jesus

Departamento de Quimica, Universidad de La Rioja, Logrono, E-26006, Spain Journal of Organic Chemistry (2005), 70(14), 5721-5724 Coobs: JOCEAH; ISSN: 0022-3263 American Chemical Society Journal SOS

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Dolence, E. Kurt; Mayer, Gabriele; Kelly, Brenda D.
School of Pharmacy, University of Myoming, Laramie, WY, 82071-3375, USA
Tetrahedron: Asymmetry (2005), 16(9), 1583-1594
CODEN: TASYE3; ISSN: 0957-4166 143:60036 Use of optically active cyclic diethyl sulfamidate 2-phosphonates chiral synthons for the synthesis of β -substituted α -amino

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AU Avenoza, Alberto; Busto, Jesus; Corzana, Francisco; Jimenez-Oses, Gonzalo; Peregrina, Jesus
CS Departamento de Quimica, Grupo de Sintesis Quimica de La Rioja, Universidad de La Rioja, Logrono, Spain Related Elements (2005), 180(5-6), 180-1859-1460
CODEN PSSLEC; ISSN: 1042-6507
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Substituted Lactams
Substituted Lactams
Milliams, Andrew J.; Charmant, Jonathan P. H.; Lawrence, Ron M.; Szetco, Peter; Gallagher, Timothy
School of Chemistry, University of Bristol, Bristol, BS8 1TS, UK
CODEN: ORLEF?; ISSN: 1523-7060
American Chemical Society
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Syntagon AB, Soedertaelje, SE-15102, Swed.
Bioorganic & Medicinal Chemistry Letters (2005), 15(6), 1637-1640
CODEN: BMCLE8; ISSN: 0960-894X
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Sun, Chaode; Bittman, Robert
Department of Chemistry and Biochemistry, Queens College, The City
University of New York, Flushing, NY, 11367-1597, USA
Journal of Organic Chemistry (2004), 69(22), 7694-7699
CODEN: JOCEAH; ISSN: 0022-3263

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Posakony, Jeffrey J.; Grierson, John R.; Tewson, Timothy J.
PET Imaging Center, Department of Radiology, University of Iowa, Iowa City, 114, 52242-1007, USA
Journal of Organic Chemistry (2002), 67(15), 5164-5169
CODEN: JOCEAH: ISSN: 0022-3263
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Williams, Andrew J.; Chakthong, Suda; Gray, Diane; Lawrence, Ron M.; Gallagher, Timothy
School of Chemistry, University of Bristol, Bristol, BS8 1TS, UK
Copanic Letters (2003), 5(6), 811-814
CODEN: ORLEF7; ISSN: 1523-7060
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Posakony, Jeffrey J.; Tewson, Timothy J.
Department of Radiology Imaging Research Laboratory, University of
Mashington, Seattle, WA, 98195, USA
Synthesis (2002), (7), 859-864
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Department of Chemistry and Biochemistry, University of Colorado, Boulder,
CO, 80309-0215, USA
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CODEN: ORLEF7; ISSN: 1523-7060
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American Chemical Society
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136:34093 136:34093 Supplication of Serine- and Threonine-Derived Cyclic Sulfamidates for the Preparation of S-Linked Glycosyl Amino Acids in Solution- and Solid-Phase Peptide Synthesis chen, Scott B.; Halcomb, Randall L. Department of Chemistry and Biochemistry, University of Colorado, Boulder, CO, 80309-0215, USA CODEN: JACSAT: ISSN: 0002-7863 American Chemical Society

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Generation of unnatural a, a-disubstituted amino acid derivatives from cyclic sulfamidates derivatives from cyclic sulfamidates derivatives from cyclic sulfamidates derivatives from cyclic sulfamidates.

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Department of Chemistry and Center for Molecular Catalysis, College of
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Efficient hydrolysis of \beta-aminosulfamic acids using a Lewis acid and a thiol for the synthesis of 2,3-diaminopropanoate derivatives

Kin, B. Moon; So, Soon Mog

Department of Chemistry and Center for Molecular Catalysis, Seoul National University, Seoul, 151-742, S. Korea
Tetrahedron Letters (1998), 39(30), 5381-5384

CODEN: TELEAY: ISSN: 0040-4039

Elsevier Science Ltd.
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Aguilera, Begona; Fernandez-Mayoralas, Alfonso
Grupo Carbolidratos, Instituto Quimica Organica, Madrid, 28006, Spain
Chemical Communications (Cambridge) (1996), (2), 127-28
CODEN: CHCOFS, ISSN-1359-7345
Royal Society of Chemistry
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Journal of Organic Chemistry (1998), 63(8), 2719-2723
CODEN: JOCEAH: ISSN: 0022-3263
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The direct synthesis of the **cyclic sulfamidate** of (S)-prolinol: simultaneous N-protection and activation towards nucleophilic displacement of oxygen Alker, David, Doyle, Kevin J.; Harwood, Laurence M.; McGregor, Andrew Pfizer Cent. Res., Sandwich/Kent, CT13 9NJ, UK Tetrahedron: Asymmetry (1990), 1(12), 877-80 CODEN: TASYE3; ISSN: 0957-4166 ANSWER 39 OF 39 CAPLUS COPYRIGHT 2006 ACS on STN 1991:143286 CAPLUS 114:143286 Tetrahedron: Asymmetry (1990), 1(12), 881-4 CODEN: TASYE3; ISSN: 0957-4166 Journal English CASREACT 114:143286 CASREACT 114:164739 Journal | | | р 1 1.23 AB L22 AN DN TI L23 AN DN SO SET SOS SE PE CS SG တ္ထ E E Fluorine for Hydroxy Substitution in Biogenic Amines: Asymmetric Synthesis and Biological Evaluation of Fluorine-18-Labeled β-Fluorophenylalkylamines as Model Systems Van Bort, Marcian E.; Jung, Yong-Woon; Sherman, Philip S.; Kilbourn, Michael R.; Wieland, Donald M. Michael R.; Wieland, Donald M. Michael R.; Wieland, Donald M. Michigan, Ann Arbor, MI, 48109-0552, USA Journal of Medicinal Chemistry (1995), 38(5), 810-15 CODEN: JMCMAR; ISSN: 0022-2623 Stereo- and regiochemical aspects of the Mitsunobu reaction in synthesis of Chiral amino ether ligands for asymmetric reactions okuda, Manabu; Tomioka, Kiyoshi Institute of Scientific and Industrial Research, Osaka University, Ibaraki, 567, Japan
Tetrahedron Letters (1994), 35(26), 4585-6
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Chem. Lab., Univ. Kent, Cantebury, CT2 7NH, UK
Journal of the Chemical Society, Chemical Communications (1993), (24), Methyl DL-3-benzyl-2,2-dioxo-1,2,3-oxathiazolidine-4-carboxylate - an intermediate for amino acid synthesis Gritsonie, Penny; Pilkington, Melanie; Wallis, John D. Chem. Lab., Univ. Kent, Canterbury, CT2 7NH, UK Acta Crystallographica, Section C: Crystal Structure Communications (1994), CSO(5), 763-701 CODEN: ACSCEE; ISSN: 0108-2701 Synthesis and stability of the cyclic sulfamidate of ANSWER 35 OF 39 CAPLUS COPYRIGHT 2006 ACS on STN 1994:591879 CAPLUS 121:191879 ANSWER 36 OF 39 CAPLUS COPYRIGHT 2006 ACS on STN 1994:579176 CAPLUS 121:179176 ANSWER 34 OF 39 CAPLUS COPYRIGHT 2006 ACS on STN 1995:380747 CAPLUS ANSWER 37 OF 39 CAPLUS COPYRIGHT 2006 ACS on STN 1994:299238 CAPLUS ANSWER 38 OF 39 CAPLUS COPYRIGHT 2006 ACS on STN 1991:164739 CAPLUS 114:164739 1857-8 CODEN: JCCCAT; ISSN: 0022-4936 American Chemical Society English CASREACT 121:179176 CASREACT 120:299238 122:182135 Journal English Journal English English Journal Journal 120: L22 AN DN TI TI LZ2 PN DN TI LA LIZZ LA LIZ L22 AN DN TI TI AU CS SEE

ANSWER 1 OF 1 CAPLUS COPYRIGHT 2006 ACS on STN

The phase transfer catalysis of PEG-X (x = 200, 400, 1000, 2000)

In epoxidn of oletins catalysed by metal porphyrin

with NaoC1 in CH2C12-H2O biphase system, a model for mimicking cytochrome by 50, was studied, and compared with that of Me(CH2)18N+Me3Br as the PTC. The exptl. results showed that PEG-X as a PTC were superior to Me3(CH2)15N+Me3Br, as the part. By anchoring PEG-X as a pTC were superior comparing the model of the past of the perphyrin, managanese tetra(p-hydrophenyl)porphyrin, metal partyric activity and stability of the Epoxidation of olefin catalyzed by metal porphyzins and polysthyles glycol. Xu, Zhenghui, Xi, Zuwesi, Jiang, Ziqi Dalian Inst. Chem. Phys., Chin. Acad. Sci., Dalian, 116012, Peop. Rep. (ACETATE OR ACETATES)
1 PORPHYRIN AND CATALY? AND SUPERIOR AND ACETATE ANSWER 1 OF 1 CAPLUS COPYRIGHT 2006 ACS on STN 1992:571098 CAPLUS 117:171098 porphyrin and cataly? and superior and acetate 3445 PORPHYRIN 24064 PORPHYRINS 40529 PORPHYRIN (PORPHYRIN OR PORPHYRINS) SUPERIOR OR SUPERIORS) China Fenzi Cuihua (1992), 6(3), 213-19 CODEN: FECUEN; ISSN: 1001-3555 11 SUPERIORS 145001 SUPERIOR 501339 ACETATE 27770 ACETATES 512670 ACETATE SUPERIOR 1289814 CATALY? 144991 Chinese => d abs

Cyclic sulfamidates: new synthetic precursors for p-functionalised a-amino acids Baldwin, Jack E.; Spive, Alan C.; Schofield, Christopher J. Byson Perrins Lab., Oxford Cent. Mol. Sci., Oxford, OX1 3QY, UK

metal **porphyrin catalyst** were greatly enhanced. The transfer of -OC from the aqueous phase to the oil phase was also facilitated. Compared with the free PEG-400 as a PTC, the PEG-400 on the metal **porphyrin** showed the better transfer ability.

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Radiolytic and photochemical reduction of carbon dioxide in solution catalyzed by transition metal complexes with some selected macrocycles Grodkowski, Jan Tech. Radiacyjnej, Inst. Chem. i Tech. Jadrowej, Warsaw, 03-195, Pol. Raporty IChTJ. Seria A (2004), 1/04, 1-56 CODEN: RISAFY; ISSN: 1425-7343 The L-number entered could not be found. To see the definition of L-numbers, enter DISPLAY HISTORY at an arrow prompt (=>). The L-number entered could not be found. To see the definition of L-numbers, enter DISPLAY HISTORY at an arrow prompt (=>). The L-number entered could not be found. To see the definition of L-numbers, enter DISPLAY HISTORY at an arrow prompt (=>). CAPLUS COPYRIGHT 2006 ACS on STN 2004:572465 CAPLUS => s 138 not 120 L39 18 L38 NOT L20 => s 139 not 121 L40 18 L39 NOT L21 => s 135 not 117 L36 18 L35 NOT L17 => s 136 not 118 L37 18 L36 NOT L18 s 130 not 112 l 18 L30 NOT L12 => s 131 not 114 L32 18 L31 NOT L14 => s 123 not 115 L33 1 L23 NOT L15 => s 132 not 115 L34 18 L32 NOT L15 18 L34 NOT L16 18 L37 NOT L19 => s 128 not 111 L29 21 L28 NOT L11 1362014 12 18 L29 NOT => d 140 1-18 ibib abs L40 ANSWER 1 OF 18 ACCESSION NUMBER: AUTHOR(S): CORPORATE SOURCE: => s 133 not 115 => s 134 not 116 => s 138 not 119 L38 NOT FOUND => s 139 not 120 DOCUMENT NUMBER: TITLE: => s 137 not 119 => s 129 not 12 L33 NOT FOUND L39 NOT FOUND 130 => s L31

DOCUMENT TYPE:

Report; General Review

ABA Review. The main goal of the work presented in this report is an explanation of the mechanism of carbon dioxide (CO2) reduction catalyzed by transition metal complexes with some selected macrocycles. The catalytic function of two electron exchange centers in the reduction of CO2, an inner metal and a macrocycle ring, was defined. Rhodium porphyrins (CIRMIIP) in alc. alkaline and slightly acidic solns. are reduced photochem, and radiolytically to RHFP—and to HRMIIP states, resp. The photocatalytic formation of HZ takes place in the system, but no porphyrins exhibit the catalytic reduction of HZ takes place in the system, but no porphyrins exhibit the catalytic reduction of RQ2. It can porphyrins in organic solvents and water are reduced photochem, and radiolytically to the FeIP-state and further reduced forms. In solns, a decay of the FeIP-state is formed via disproportionation of the FeIP-state. The latter secondaries lead to the formation of CO. HCO2—and HZ. Application of peterphenyl (TP) as an addh. Horosensitizer increases by one order of magnitude the yield of CO and HCO2—formed during photolysis of CO2—stricthylamine to MUP2—states. The Macrose sphotocred, to the radical anion TP —.

The latter, species as a strong reductant, reduces subsequently metalloporphyrins to MUP2—states. The MMP2—state of a porphyrin is responsible for CO2 reduction Side reactions lead to the formation of the porphyrins, are characterized by the more extended anomatic structure and they are more resistant for degradation COsalt and iron phthalocyanines are easily reduced to the MUP2—state. However, the latter state does not resistant for degradation colonial cornins show, that these complexes are more efficient in the catalysis of CO3 in comparison with porphyrins. Investigations of CO3 in comparison with porphyrins. Investigations of CO3 in comparison with cobalt porphyrins. Poly substitution of corned water at higher oxidation states in complex stabiles are easily endiced to made a more released by complexes and metal cente CO2 and thus it becomes a precursor of CO and HCOO- formation, with the yields comparable with metalloporphyrin systems. Due to the fact, that side reactions cause a ring degradation, it was checked whether the metal ions released during the degradation process could also catalyze CO2 reduction It porphyrins. CO2 and thus

found that FeI iron ions react with CO2 to form an adduct, a direct precursor of CO. A protonated form of FeI is responsible for H2 formation instead. A possibility of four., six- and eight-electrons CO2 reduction was studied in aqueous solns. containing CO2 and CuII copper ions. This study was initiated by the observation of methane formation during electrochem. reduction of CO2 on the copper electrode. It was established that a presence of CuI ions and a reduced form of CO2-radical anion "CO2-, is

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ACCESSION NUMBER: DOCUMENT NUMBER: TITLE:

2003:485180 CAPLUS 139:187329

Reversible Electrochemical Generation of a Rhodium(II) Porphyrin: Thwarting Disproportionation with Weakly Coordinating Anions Sun, Haoran; Xue, Feng; Nelson, Andrew P.;

Redepenning, Jody; DiMagno, Stephen G.
Department of Chemistry and Center for Materials
Research and Analysis, University of Nebraska,
Lincoln, NE, 68588-0304, USA
Inorganic Chemistry (2003), 42(15), 4507-4509

CORPORATE SOURCE:

AUTHOR (S):

CODEN: INOCAJ; ISSN: 0020-1669 American Chemical Society Journal DOCUMENT TYPE: PUBLISHER:

English

AB The authors report electrochem, generation of a stable Rh(II) porphyrin (RNII(F28PPP)) from a 4-coordinate Rh(II) precursor (RNI(F28PPP)]-dissolved in weakly coordinating electrolyte solns. This work provides the 1st example of an unambiguously reversible 1-electron electrochem.

oxidation of a RNI(por), and demonstrates that electrochem.

oxidation can be performed under conditions that are compatible with alkane activation. These studies begin to classify those media capable of supporting a stable RNI(por), and those that induce disproportionation.

REFERENCE COUNT: THERE ARE 21 CITED REFERENCES AVALLABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT LANGUAGE:

COPYRIGHT 2006 ACS on STN 2003:286777 CAPLUS CAPLUS L40 ANSWER 3 OF 18 ACCESSION NUMBER:

138:411108 DOCUMENT NUMBER:

Synthesis and Properties of Rhodium(III) Porphyrin Cyclic Tetramer and Cofacial Dimer

Fukushima, Keiko; Funatsu, Kenji; Ichimura, Akio; Sasaki, Yoichi; Suzuki, Masamitsu; Fujihara, Tetsuaki; Tsuge, Kiyoshi; Imamura, Taira

AUTHOR(S):

CORPORATE SOURCE:

Division of Chemistry, Graduate School of Science, Hokkaido University, Sapporo, 060-0810, Japan Chorganic Chemistry (2003), 42(10), 3187-3193 CODER: INOCAU, ISSN: 0020-1669

SOURCE:

American Chemical Society Journal English DOCUMENT TYPE: PUBLISHER: LANGUAGE:

CASREACT 138:411108 OTHER SOURCE(S):

AB Rhodium(III) porphytin complexes, (Rh(4-PyT3P)Cl]4 (1) and
Rh (2-PyTB3P)Cl]2 (2) (4-PyT3P = 5-(4-PyTidyl)-10.15,20tritolylporphytinato dianion, 2-PytB3P = 5-(4-PyTidyl)-10.15,20tritolylporphytinato dianion, 2-PytB3P = 5-(2-PyTidyl)-10.15,20tritolylporphytinato dianion, were self-assembled and characterized
butyl)phenylporphytinato dianion, were self-assembled and characterized
butyl)phenylporphytinato dianion, were self-assembled and characterized
by IH NMR spectroscopy, IR spectroscopy, and electron spray
ionization-mass spectroscopy methods. The spectroscopic results certified
that the rhodium porphyth complexes I and 2 have a
cyclic tetrameric structure and a cofacial dimeric structure, resp. The
x-ray structure and. of I confirmed the cyclic structure of the complex.
The Soret bands of both oligomers were significantly broadened by
excitonic interactions between the porphyth units, compared to those
observed for a corresponding analog of Rh(TPP) (PyCl) (TTP =
5,10,15,20-tetratolylporphytinato dianion, Py = pyridine). Stepwise
oxidation of the porphytin rings in the oligomers was observed by cyclic
voltammetry. The oligomers land 2 are very stable in solution, and they
slowly undergo reactions with pyridine to give corresponding monomer
complexes only at high temps. (apprx 180?)

REFERENCE COUNT:
23 THERE ARE 23 CIFED REFERENCES AVAILABLE FORMAT

REFERENCE COUNT:

COPYRIGHT 2006 ACS on STN L40 ANSWER 4 OF 18 CAPLUS ACCESSION NUMBER:

2002:189892 CAPLUS
Electrochemistry of a series of rhodium
Electrochemistry of a series of the
Electrochemical properties and
relationship between electrochemical properties and

porphyrins DiMagno, Stephen G.; Sun, Haoran; Biffinger, Justin; Nelson, Andrew P. coordination properties of the rhodium

CORPORATE SOURCE:

SOURCE:

AUTHOR(S):

Nebraska-Lincoln, Lincoln, NE, 68588-0304, USA Abstracts of Papers, 223rd ACS National Meeting, Johando, El, United States, April 7-11, 2002 (2002), INOR-289. American Chemical Society: Washington, D. Department of Chemistry, University of

cyclic voltammetry, square wave voltammetry, macroelectrode steady voltammetry, in-situ spectroelectrochem, and digital simulation in various media. The electrochem, properties are strongly dependent on the coordination properties of the thodium porphyrins.

For example, when triphenylphosphine presents in 1,2-difluorobenzene solution two steps reversible one-electron oxidation of thodium(1) thiaporphyrin instead of one step irreversible two-electron oxidation of perfluorinated rhodium(1) porphyrin are observed. The detailed electrochem reaction mechanism and the relationship between electrochem. porperties and coordination properties of the rhodium. The electrochem. properties of a series of rhodium thiaporphyrins and perfluorinated rhodium porphyrins are investigated by Conference; Meeting Abstract C. CODEN: 69CKQP English DOCUMENT TYPE: LANGUAGE:

Electrochemical generation of rhodium porphyrin hydrides. Catalysis of hydrogen evolution US COPYRIGHT 2006 ACS on STN 1997:483409 CAPLUS 127:168078 ANSWER 5 OF 18 CAPLUS ACCESSION NUMBER: DOCUMENT NUMBER:

Grass, Valerie; Lexa, Doris; Saveant, Jean-Michel Labocatoire d'Electrochimie Moleculaire de l'Universite Denis Diderot, Unite Associee au CNRS No 438, Paris, 75251, Fr. Journal of the American Chemical Society (1997), 119(32), 7526-7532 CODEN: JACSAT; ISSN: 0002-7863 AUTHOR(S): CORPORATE SOURCE: SOURCE:

American Chemical Society English Journal DOCUMENT TYPE: PUBLISHER:

AB In polar solvents, Rhilli porphyrins are directly reduced in Rhill complexes which react readily with Bronsted acids to give Rhilli complexes which react readily with Bronsted acids to give Rhilli hydrides. They then undergo, at a more neg. potential, an addnl. electron uptake to yield the corresponding Rhill hydrides. The electrogenerated rhodiunlil complex is the key intermediate of catalytic hydrogen evolution according to a mechanism which heavily depends on the solvent and on axial ligands. In DMSO, hydride transfer from Rhill H- to the acid, yielding H2, competes with hydride transfer from Rhill H- to the butyronitrile, hydrogen evolution occurs both by hydride transfer to the acid and H-atom abstraction to the solvent. The latter pathway is shut of by the addition of strong and soft ligands such as tertiary phosphines. With PEE3, a particularly strong electron-donating ligand, not only Rhill H- but also Rhill H riggers H2 evolution. The various changes of the hydrogen evolution mechanism as well as the stability of the catalyst can be reationalized by the variation of the electron distribution brought about by the presence of the absence of the axial ligand. REFERENCE COUNT: LANGUAGE: AB In po

JUS COPYRIGHT 2006 ACS on STN 126:32342 Reductive Electrochemistry of Rhodium L40 ANSWER 6 OF 18 CAPLUS ACCESSION NUMBER: 199 DOCUMENT NUMBER:

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

Porphyrins. Disproportionation of Intermediary Oxidation States

Saveant, Jean-Michel Laboratorie d'Electrochimie Moleculaire, Universite Danis Diderct, Paris, 78551, Fr. Journal of the American Chemical Society (1997), Grass, Valerie; Lexa, Doris; Momenteau, Michel; CORPORATE SOURCE:

SOURCE:

AMOUNCE the injusts of the injusts of the metal metal of the injusts of this irreversible reaction yielding directly the Rh(I) complex. The cause of this irreversiblity is not the metal-metal dimerization of the cause of this irreversiblity is not the metal-metal dimerization of the injust of the injust of the injustance of this irreversiblity is not the metal-metal dimerization of the injust of sarting Rh(II) complex as believed earlier but rather deligation which generates a secondary Rh(II) species easier to reduce than the starting Rh(III) popphyrin, such as those bearing cross-trans basket-handle superstructures which forbid the approach of two mols. at bonding distance, exhibit the same benavior as simple rhodium porphyrins. The occurrence of such an ECE-disproportionation process, seldom observed in the redox chemical of metalloporphyrins or similar complexes, is probably related to the tendency of the rhodium atom to shift out of the porphyrin plane, particularly at the Rh(I) oxidn state. It is remarkable that strong and soft ligands, e.g., tertiary phosphines, annihilate the disproportionation of the rhodium(III) complex.

Serent of the porphyrin plane of the Rh(III) or the REFERENCE COUNT:

Serent of the porphyrin plane of the Rh(II) or the REFERENCE COUNT:

Serent of the porphyrin plane of the Rh(II) or the REFERENCE COUNT: 119(15), 3536-3542 CODEN: JACSAT; ISSN: 0002-7863 American Chemical Society English Journal REFERENCE COUNT: DOCUMENT TYPE: LANGUAGE:

cyclopropanation of olefins by diazo esters
Brown, Kathlynn Corinne
Univ. of Texas, Austin, TX, USA
(1994) 189 pp. Avail.: Univ. Microfilms Int., Order
No. DA9519254 ACCESSION NUMBER: 1995;775551 CAPLUS
DOCUMENT NUMBER: 123:228440
TITLE: Novel chemistry of highly reactive metal complexes: oxidative cross-linking of proteins mediated by a nickel-peptide complex and investigations of the rhodium porphytin-catalyzed
rhodium porphytin-catalyzed AUTHOR(S): CORPORATE SOURCE: SOURCE:

From: Diss. Abstr. Int., B 1995, 56(2), 821

Dissertation English CAPLUS COPYRIGHT 2006 ACS on STN L40 ANSWER 8 OF 18 ACCESSION NUMBER:

Unavailable

LANGUAGE: AB Unave

DOCUMENT TYPE:

Factors influencing the site of electroreduction in 1993:436225 CAPLUS 119:36225 DOCUMENT NUMBER AUTHOR(S): TITLE:

rhodium porphyrins
Kadish, K. M.; Hu, Y.; Tagliatesta, P.; Boschi, T.
Dep. Chem., Univ. Houston, Houston, TX, 77204-5641, Inorganic Chemistry (1993), 32(14), 2996-3002 CODEN: INOCAJ; ISSN: 0020-1669 CORPORATE SOURCE: SOURCE:

English DOCUMENT TYPE: LANGUAGE:

The electrochem. of Milli porphyrins containing bound phosphine, isocyanide, or carbene axial ligands was studied by cyclic voltammetry and UV-visible spectroelectrochem. In THF and CH2C12 containing Bu4MPF6 as supporting electrolyte. The studied compds. are represented as [(TPP)RR[L]2]PF6, (TPP)Rh[L']PF6, or (TPP)Rh[PF3] (OH), where TPP is the dianion of tetrapherylporphyrin, L = PPH3, PPHZMe, PPHMe2, and CKCH2PH, and L' = (CKHKH2Ph)2. The addition of I electron to these complexes leads to 1 of 2 different reduction products, depending upon the temperature and the specific

axial ligands. Some of the complexes are reversibly reduced by 1 electron of yea a transient Rh[III] porphyrin π anion radical, while others are irreversibly reduced under the same solution conditions to give dimeric [(FPP)Rh]2. In several cases, the addition of 1 electron gives a Rh[II] set of

dimer at room temperature but a Rh(III) π anion radical at low temperature The

initially reduced at the porphyrin π ring system, and this is also the conclusion based on electrochem. criteria relating the potentials for **oxidation** and reduction of each metalloporphyrin in nonaq. media. The absolute p.d. between E1/2 for the lst room temperature **oxidation** of a given complex in CH2C12 and the lst low-temperature reduction of the same species in UV-visible data suggest that all of the studied Rh(III) porphyrins are

THE

(the reaction is reversible) ranges at 2.22-2.32 V, suggesting that both electrode reactions involve the porphyrin π ring system. One of the species, (TPP)Rh(FP3) (OH), undergoes a slow conversation of the electrogenerated π anion radical to dimeric ([PPP)Rh]2, and this reaction was followed as a function of time by thin-layer UV-visible spectroelectrochem. In THF. Exchange equilibrium involving bound PPH3 and THF axial ligands were also studied in CH2CL2 or THF by UV-visible spectroscopy. Both [(TPP)Rh(PPH3)]+ and [(TPP)Rh(PPH3)]+ are converted to ([TPP]Rh(PPH3)[THF]+ in near THF, but the addition of 1.0 equiv of PPH3 to these solns. leads to [(TPP)Rh(PPH3)]+ as dentified by UV-visible spectroscopy. The formation constant for this reaction was calculated as 103.1 using spectrophotometric methods.

JUS COPYRIGHT 2006 ACS on STN 1992:407531 CAPLUS 117:7531 L40 ANSWER 9 OF 18 CAPLUS ACCESSION NUMBER: 199

DOCUMENT NUMBER:

Asymmetric cyclopropanation of alkenes catalyzed by a rhodium chiral fortress porphyrin O'Malley, Sean; Kodadek, Thomas AUTHOR (S):

Dep. Chem. Biochem., Univ. Texas, Austin, TX, 78712, USA CORPORATE SOURCE:

Organometallics (1992), 11(6), 2299-302 CODEN: ORGND7, ISSN: 0276-7333

Journal

SOURCE:

DOCUMENT TYPE:

English

Khe synthesis and catalylic cyclopropanation activity of a new porphyrin known as the chiral fortress macrocycle is reported. This mol. has optically pure naphthyl-pyrenyl groups appended directly to the meso carbons of the porphyrin. The iodorhodium derivative is a catalyst for the cyclopropanation of alkenes by Et diazoacetate. The syn cyclopropyl esters are the major product in each case examined except one. In some cases very high diastereoselectivity is observed The enantiomeric excess resulting from chiral fortress-mediated reactions are modest.

ANSWER 10 OF 18 CAPLUS COPYRIGHT 2006 ACS on STN SSION NUMBER: 1989:477131 CAPLUS L40 ANSWER 10 OF ACCESSION NUMBER: DOCUMENT NUMBER:

111:77131 Synthesis and applications of metalloporphyrins. Catalytic reactions by **rhodium** porphyrins

AUTHOR(S): CORPORATE SOURCE:

Ögoshi, Hisanobu Fac. Eng., Kyotc Univ., Kyotc, 606, Japan Yuki Gosei Kagaku Kenkyusho Koenshu (1989), 3, 23-31 CODEN: YGKKEI; ISSN: 0913-8463 Journal; General Review DOCUMENT TYPE: SOURCE:

LANGUAGE: AB A rev

A review with 5 refs., on catalytic reactions using rhodium porphyrin complexes, especially oxidation of olefins, reduction of ketones, and aldol condensations.

L40 ANSWER 11 OF ACCESSION NUMBER: DOCUMENT NUMBER: TITLE:

ANSWER 11 OF 18 CAPLUS COPYRIGHT 2006 ACS on STN
SSION NUMBER: 1988:428951 CAPLUS
MENT NUMBER: 109:28951
E: Electrochemical and spectroscopic studies of
(P)RM/R() (1) and [(P)Rh(L)2]+ where P is the dianion of octaethyl— or tetraphenylporphyrin, R is a

The electrochem. and spectroelectrochem. of (P)Rh(R), (P)Rh(R)(L), and (P)Rh(R)(L), where P is the diamion of octaethylporphyrin (OPP) or tetraphenylporphyrin (TPP), R is CH3, CZH5, or C4H9, and L is triphenylporphyrin (TPP), R is CH3, CZH5, or C4H9, and L is triphenylporphyrin (PPP), R is CH3, CZH5, or C4H9, and L is triphenylporsphine or dimethylphosphine are reported. At polarog. confors. of (P)Rh(R) (apprx.10-3 M), the binding of 1 triphenylphosphine ligand and the formation of (P)Rh(R)(RPM3) were observed This contrasts to lower porphyrin comens. where the bis(triphenylphosphine) adduct (P)Rh(R) (PPM3) and resolvent (P)Rh(R) (PPM3) were calculated by using electrochem. and spectroscopic methodologies and varied between 1.0 and 4.0 + 103 depending upon the porphyrin macrocycle (OEP or TPP), the specific R group, and the solvent (methylene calculated or benzonitrile). The electrocedn. of (P)Rh(R)(PPM3) initially leads to a porphyrin n-anion radical and the transient Commation of (P)Rh(R)(PPM4)]. This reaction was characterized by thin-layer electrochem. time scale. The formation of (PPR) (PPRM6)(PPM4)]. This reaction was characterized by thin-layer spectrached. and provides the lst example for reduction of a non-σ-bonded Rh(III) porphyrin at the porphyrin n-ring system. o-bonded alkyl group, and L is triphenylphosphine or dimethylphenylphosphine Kadish, K. M.; Araullo, C.; Yao, C. L. Dep. Chem., Univ. Houston, Houston, TX, 77004, USA COGRIN ORGND7; ISSN: 0276-7333 English CORPORATE SOURCE: DOCUMENT TYPE: AUTHOR (S): SOURCE:

1988:413536 CAPLUS L40 ANSWER 12 OF 18 CAPLUS ACCESSION NUMBER: 1988

DOCUMENT NUMBER:

Electrochemical studies of dimeric rhodium(III) porphyrins containing a dibasic nitrogen-heterocyclic bridging ligand a dibasic nitrogen-heterocyclic liu, Y. H.; Anderson, J. E.; Kadish, K. M. Dep. Chem., Univ. Houston, TX, 77004, USA Inorganic Chemistry (1988), 27 (13), 2320-5 CODEN: INOCAJ; ISSN: 0020-1669 109:13536

CORPORATE SOURCE: AUTHOR (S):

Journal English DOCUMENT TYPE: SOURCE:

The electrochem. and spectroelectrochem. of [(P)RhCl]2L, where P is the diamion of tetraphenylporphyrin (TPP) or octaethylporphyrin (OEP) and L is a conjugated dibasic N-heterocyclic ligand such as 4.4-blpyridine (bpy), trans-1,2-bis(4-pyridyl)ethylene (BPE) or a nonconjugated N-heterocyclic ligand such as 1,2-bis(4-pyridyl)ethane (BPE) or 4,4'-Lipyridine (bpy), trimetylenebis(pyridine) (TMDP), are reported. The Rh(III) dimers with BPA or TMDP nonconjugated bridging ligands undergo i irreversible metal-centered reduction in THF or methylene chloride. However, 2 overlapping irreversible metal center redns. are observed for Rh(III) dimers that are linked via the conjugated bridging ligands, bpy and BPE. In all cases, [(P)Rh)2 and the free N-heterocyclic ligand are generated as products from a more chemical reactions that follow the metal-centered reduction of

Rh(III)

to Rh(II). Two reversible 2-electron **oxidns**. are observed for (P)RhCl]2Ly, where L = BPE, BPA, and TMDP. This behavior contrasts with the case for [(P)RhCl]2Dpy, which undergoes a single reversible 2-electron transfer followed by 2 reversible 1-electron **oxidns**. On the basis of the electrochem. and spectroelectrochem. data, an overall mechanism for reduction and ${\bf oxidation}$ of the [(P)RhCl]2 complexes is presented.

CAPLUS COPYRIGHT 2006 ACS on STN 1988:112169 CAPLUS 108:112169 L40 ANSWER 13 OF 18 ACCESSION NUMBER: DOCUMENT NUMBER: TITLE:

Photosensitized hydride transfer. Highly

regioselective 1,4-photoreduction of NAD(P)+ models Dep. Mater. Sci. Technol., Technol. Univ. Nagaoka, Niigata, 940-21, Japan (Chemistry Letters (1987), (8), 1651-4 CODEN: CMLTAG, ISSN: 0366-7022 under visible light with an organometallic rhodium(III) porphyrin as sensitizer Aoyama, Yasuhiro; Midorikawa, Koji; Toi, Hiroo; Ogoshi, Hisanobu Journal English CASREACT 108:112169 DOCUMENT TYPE: LANGUAGE: OTHER SOURCE(S): CORPORATE SOURCE: AUTHOR (S): SOURCE:

On irradiation with visible light the Rh porphyrin complex I catalyzed the reduction of pyridinium ions II (R = CONH2, AC) by Rh8— with Ma2CHOH as a proton source to give the 1,4-dihydropyridines III as the sole reduction products and biphenyl as the **oxidation** product of Ph4B—. æ

П

Univ. Houston, Houston, TX, USA (1987) 275 pp. Avail.: Univ. Microfilms Int., Order No. DA8714775 br. Diss. Abstr. Int. B 1987, 48(4), 1029-30 Dissertation CAPLUS COPYRIGHT 2006 ACS on STN 1987:624874 CAPLUS 107:224874 Electrochemistry of rhodium fao, Chaoliang porphyrins English ANSWER 14 OF 18 L40 ANSWER 14 OF ACCESSION NUMBER: DOCUMENT NUMBER: AUTHOR(S): CORPORATE SOURCE: SOURCE: Unavailable DOCUMENT TYPE: LANGUAGE: g

CAPLUS COPYRIGHT 2006 ACS on STN 1986:108822 CAPLUS 104:108822 CAPLUS CAPLUS CATALLOPORPHYRINS. 1. CATALLYTIC modification of borane reduction of ketone Catalytic modification of borane reduction of ketone L40 ANSWER 15 OF 18 ACCESSION NUMBER: DOCUMENT NUMBER:

with rhodium(III) porphyrin as catalyst Aoyama, Yasuhiro; Fujisawa, Takamichi; Toi, Hiroo; Ogoshi, Hisanobu Dep. Mat. Sci., Technol. Univ. Nagaoka, Nijgata, 949-54, Japan Journal of the American Chemical Society (1986), CORPORATE SOURCE:

CODEN: JACSAT; ISSN: 0002-7863 108(5), 943-

Journal DOCUMENT TYPE:

UAGE: $(Octaethyl-\ or\ tetraphenylporphyrinato)\ rhodium(III)\ chloride\ shows\ an$ LANGUAGE: AB (Octa

efficient catalysis in the aerobic reduction of ketone with NaBH4 in THF. I initial step in the catalytic cycle is the rate-determining complexation of BH4-

borane transfer from the adduct to ketone to give dialkoxyborane and hydridorhodium species. In the subsequent step, the Rh-H species undergoes oxidation with O2 back to RhIII with concomitant hydrolysis of dialkoxyborane to alc. Essentially, autorecycling RhIII and Rh-H act as a borane generator and proton source, resp., in a catalytic manner. Furthermore, the RhII-BH4 complex capable of transferring borane to ketone lacks what is characteristic of free borane, i.e., facile oxidation with O2 and ready hydrolysis with H2O. Thus, the present system provides a highly efficient, catalytic modification of synthetic reactions of borane in the presence of oxygen. with RhIII porphyrin (RhIII + BH4- → RhIII-BH4) followed by a rapid

L40 ANSWER 16 OF 18 CACCESSION NUMBER: DOCUMENT NUMBER: TITLE:

CAPLUS COPYRIGHT 2006 ACS on STN
1984:414298 CAPLUS
101:14298
Optical absorption and ESR spectra of monomeric
rhodium(II) tetraphenylporphyrin in
2-methyletetrahydrofuran solution at 77 K
Hoshino, Mikko, Yasufuku, Katsutoshi, Konishi, Shiro;
Imamura, Masashi

Solar Energy Res. Group, Inst. Phys. Chem. Res., Wako, 351, Japan Inorganic Chemistry (1984), 23(13), 1982-4 CODEN: INOCAJ; ISSN: 0020-1669

CORPORATE SOURCE:

SOURCE:

AUTHOR(S):

Journal DOCUMENT TYPE:

Rh porphyrins have aroused much attention because of the wide variety of their chemical reactions. For instance, Rh porphyrins reachly react with the simple mols. H2. 02, and No to produce hydride, oxygen, and nitric adducts, resp. The Rh atom incorporated in porphyrin ligands is known to have 3 oxidation states, +1, +2, and +3. However, monomeric Rh(II) because of their propensity to facile dimerization. This note reports the optical absorption and ESR spectra of monomeric Rh(II) tetraphenylporphyrin (RhIITPPP) produced by photolysis of chloro(tetraphenylporphinato) rhodium(III) (ClhhIITPPP) and dimeric RhIITPP ((RhIITPPP) in 2-methyltetrahydrofuran solns. at 77 K.

PA, PLUS COPYRIGHT 2006 ACS on STN 1981:50764 CAPLUS 95:10764 Dioxygen and nitric oxide complexes of **rhodium** Wayland, Bradford B.; Newman, Alan R. Dep. Chem., Univ. Pennsylvania, Philadelphia, 19104, USA porphyrins CAPLUS ANSWER 17 OF 18 ACCESSION NUMBER: AUTHOR(S): CORPORATE SOURCE: DOCUMENT NUMBER:

Inorganic Chemistry (1981), 20(9), 3093-7 CODEN: INOCAJ: ISSN: 0020-1669 Journal English DOCUMENT TYPE: SOURCE:

(RhOEP)2 (H2OEP = octaethylporphine) reacts with O to form RhOEP(02) (S = 1/2), which subsequently forms the µ-percox complex (RhOEP)202. EPR studies of RhOEP(2) and RhTPP(02) (H2TPP = tetraphenylporphine) and their 1:1 donor complexes are reported and compared with those of the Co analogs. (RhOEP)2, RhOEP(H), and RhOEP(CI) all react with NO to ultimately product the same product, RhOEP(ND). The reactions of RhOEP(CI) and RhTPP(CI) with NO proceed through a metastable paramagnetic intermediate Rh(por)(CI)(NO) (por = porphine) which from EPR and electronic spectral studies is formulated as containing a porphyrin recation radical unit with an 2Alu ground state. RhOEP(NO)(CI) assocs. to form a radical dimer (S = 1) with D = 5.17 + 10-3 cm-1, E = 2.4 + 10-4 cm-1, but only monomeric RhTPP(NO)(CI) is observed Electrochem. LANGUAGE: AB (RhOI

studies of Rh(III) porphyrins also support the porphyrin cation radical formulation for Rh(por)(NO)(Cl) complexes.

ACCESSION NUMBER: 1973:111487 CAPLUS
DOCUMENT NUMBER: 1973:111487 CAPLUS
DOCUMENT NUMBER: 78:111487

TITLE: New thodium(I)-porphyrin complex. II. Synthesis and oxidative alkylation complex. II. Synthesis and oxidative alkylation complex. II. Synthesis and oxidative alkylation complex. II. Synthesis and Ogoshi, H.; Omura, T.; Yoshida, Z. CORPORATE SOURCE: Dep. Synth. Chem., Kyoto Univ., Yoshida, Japan Journal of the American Chemical Society (1973), 95(5), 1666-8

CODEN: JACSAT; ISSN: 0002-7863
DOCUMENT TYPE: English

DOCUMENT TYPE: Journal John William (I) Document TYPE: Document TYPE: Document TYPE: English

R Treating N-methyloctaethylporphyrin with di-µchorobis[dicarbonylihodium [I]] gave a rhodium [I] porphyrin complex
containing I N-methylporphyrin and one [Rh(CO)2CI]2 and formulated as
[N-methylporphyrin] [Rh(CO)2CI]2 [I]. Spectral data indicate N-H and N-Me
bonds on the porphyrin inner core. The 220 Miz NMR spectrum of I shows
quite low symmetry for the porphyrin frame. I is oxidized to a
mono-Rh(III) porphyrin complex with loss of the N-Me3 bond and methylation
of the Rh(III) atom.

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